

Transition Metal- Catalyzed Asymmetric Conjugate Addition

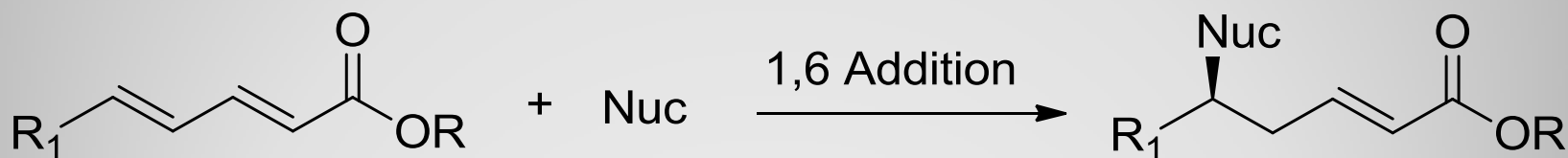
Rachel Whittaker
Dong Group
September 26, 2012

- Introduction
 - Background
 - History
- First Row TM Catalyzed
 - Copper
 - Nickel
- Second Row TM Catalyzed
 - Rhodium
 - Palladium
- Conclusions

Overview

What is Conjugate Addition?

- "Addition of a nucleophile to activated alkenes or alkynes, followed by protonation of the anionic intermediate."



1,2 vs 1,4

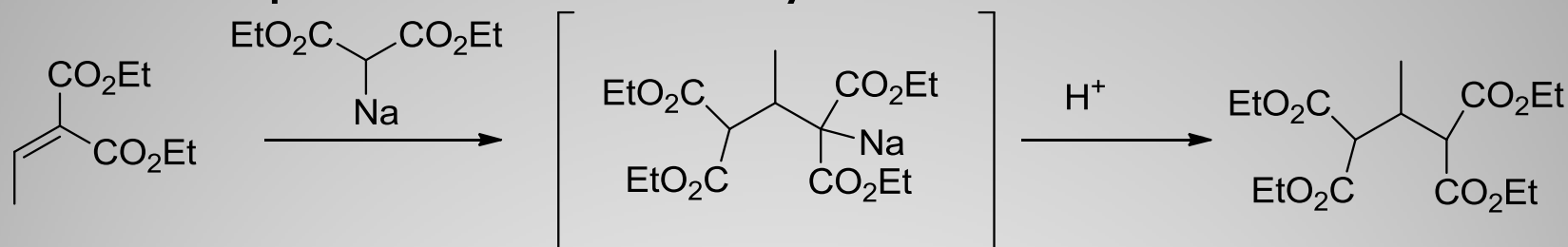
- Ways to Favor 1,4 Over 1,2:
 - Soft nucleophiles
 - Polar solvents (HMPA)
 - Sterics
 - Delocalization of carbanion
 - Higher temperatures
- With catalysts, tuning of structure/ligands also affects this

Why Asymmetric Conjugate Addition?

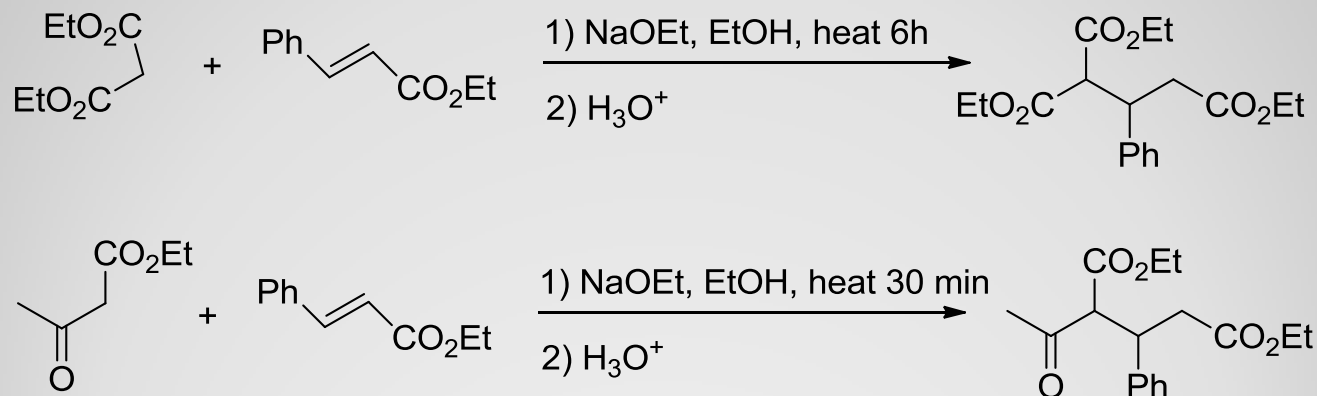
- Powerful method of forming C-C bonds with a new stereocenter
- Enantiopure compounds important in industry (pharmaceuticals and materials), as well as academia
- Tandem reactions lead to new functionality in surprising and useful ways

History

- First reported in 1883 by Komnenos



- Michael Reaction



Komnenos, T., *Liebigs Ann. Chem.*, **1883**, 35, 145.
Michael, A., *J. Prakt. Chem.*, **1887**, 3, 349.

Development and Advances

- Field has developed quickly due to need of C-C bond formation and stays relevant
- Asymmetric conjugate addition still relatively young field
- Split into two broad sections:
 - Transition metal catalysts (esp. Cu and Rh)
 - Organocatalysts (commonly a heterocycle)

First Row Transition Metals

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	35 Br	36 Kr				
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	53 I	54 Xe				
55 Cs	56 Ba	Ac		72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	(117)	(118)			
(119)	(120)	(121)	(154)											(167)	(168)		

LANTHANIDES

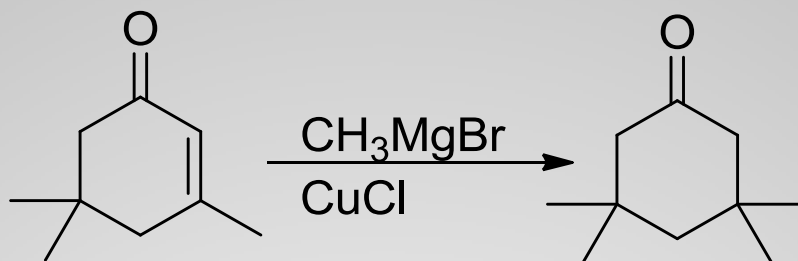
58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
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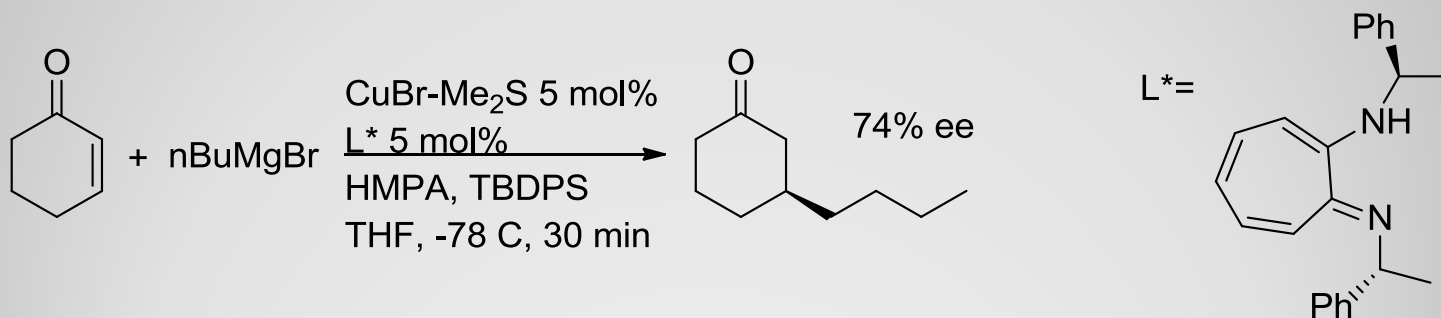
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
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History

- Kharasch group discovered trace amounts of Cu salts led to 1,4 addition



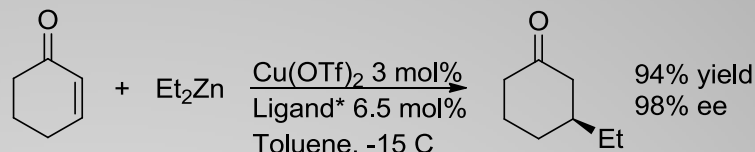
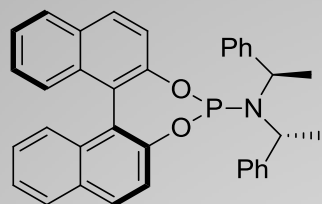
- Lippard was the first enantioselective example



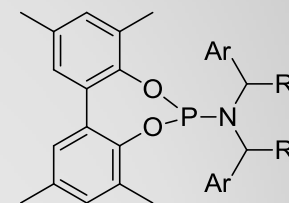
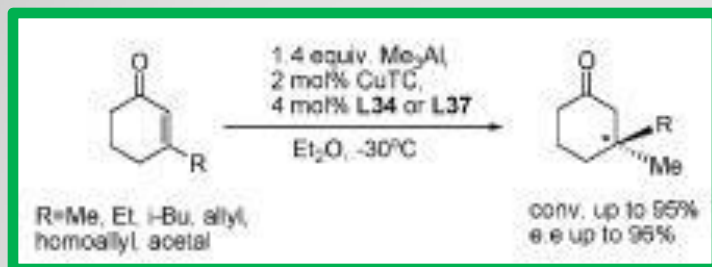
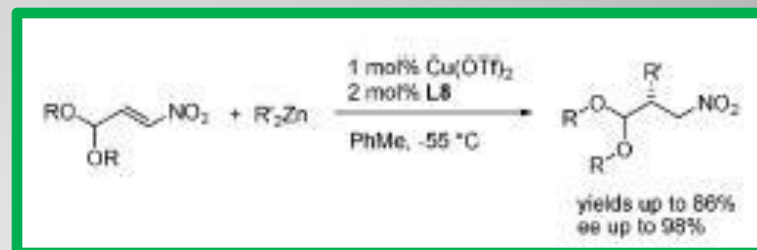
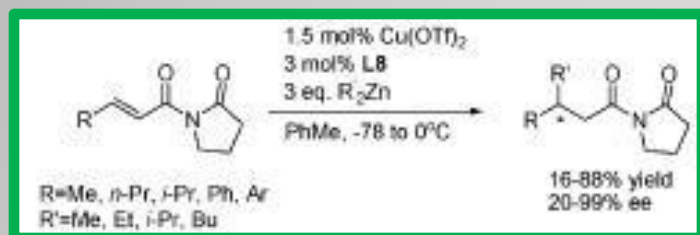
Kharasch, M.S., Tawney, P.O., *J. Amer. Chem. Soc.*, **1941**, 63, 2308.

Lippard, S.J., *Organometallics*, **1990**, 9, 3178.

Organozinc/Organoaluminum

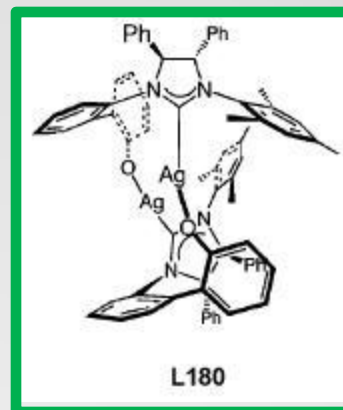
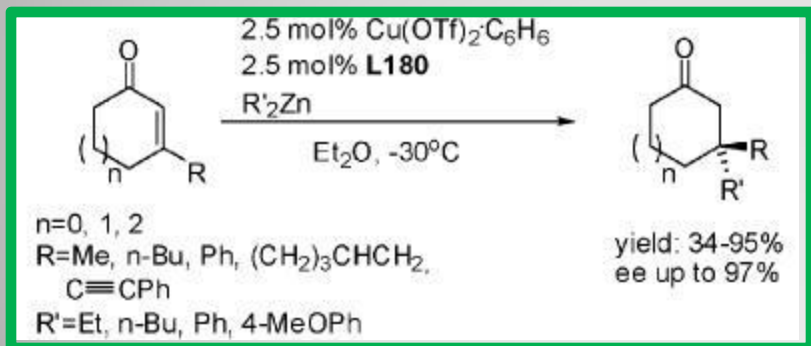
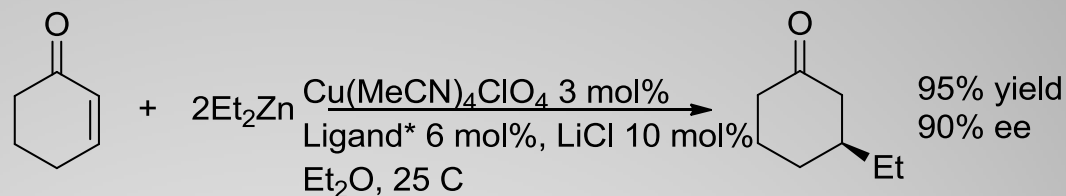
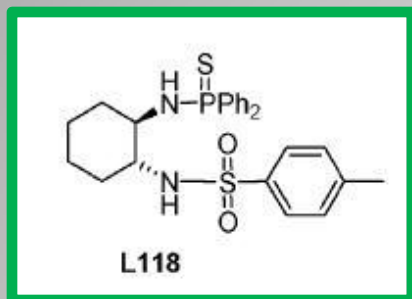


Me, Hep, iPr, Ph(CH₂)₃ also showed good ee's

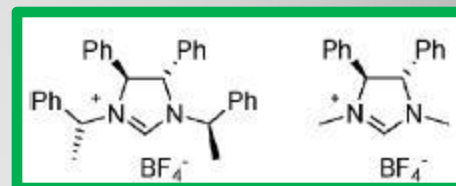


Ar= Ph, 2-Naph
R=(S)-Me, (R)-Me

More Organozinc Ligands

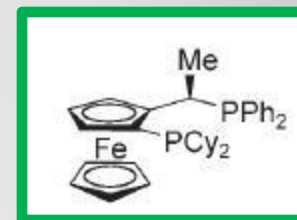
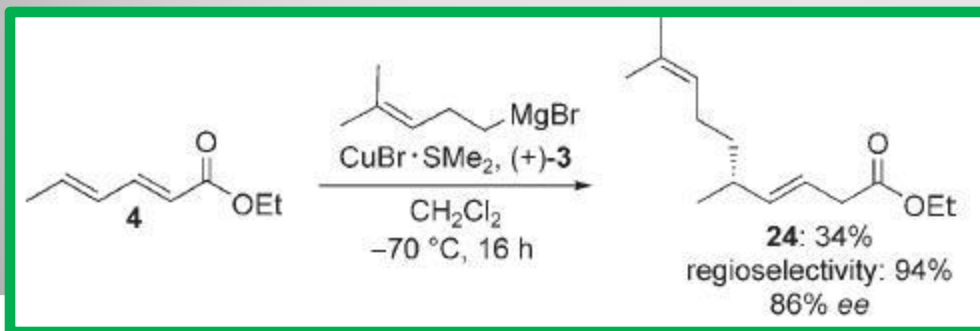
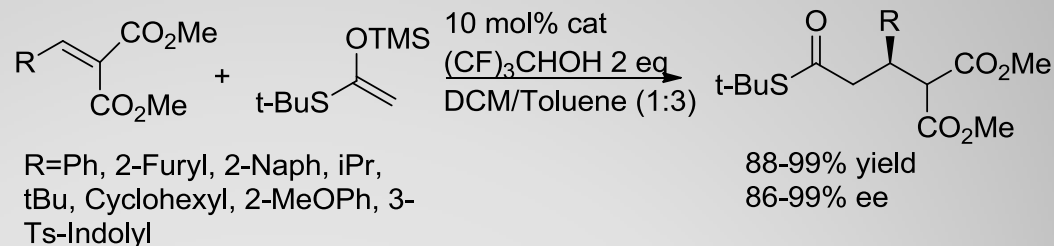
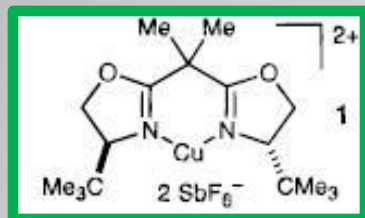
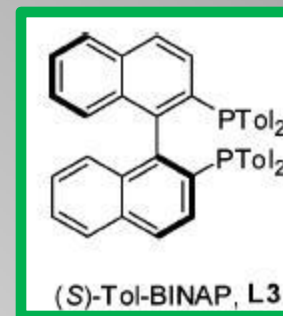
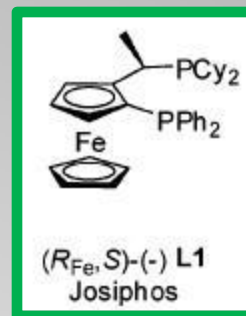
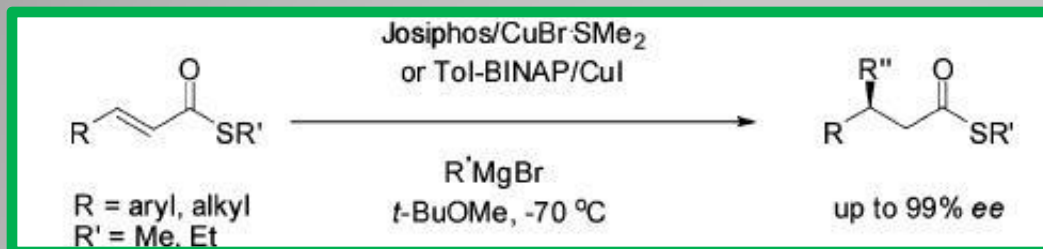


Other NHC Ligands



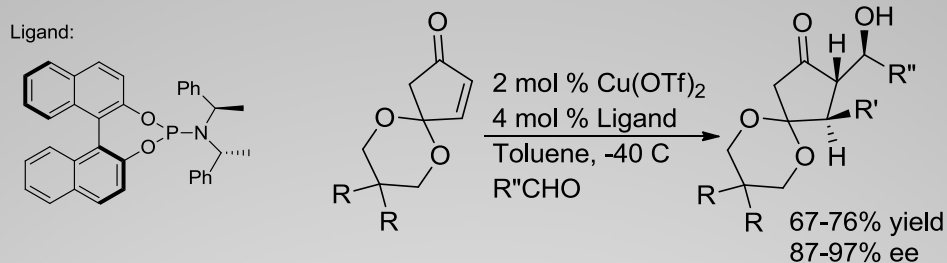
Shi, M., *Adv. Synth. Catal.*, **2005**, 347, 535.
 Hoveyda, A.H., *J. Am. Chem. Soc.*, **2006**, 128, 7182.

Other Cu Reagents

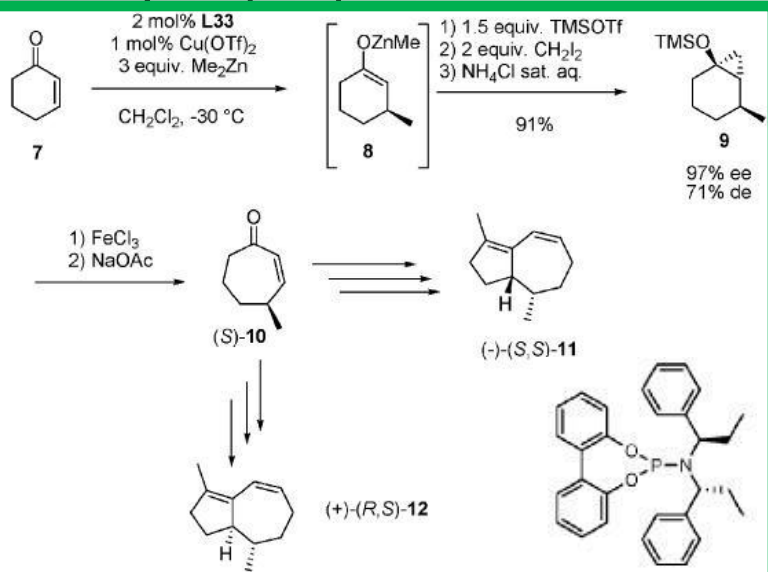


Applications: Tandem Reactions

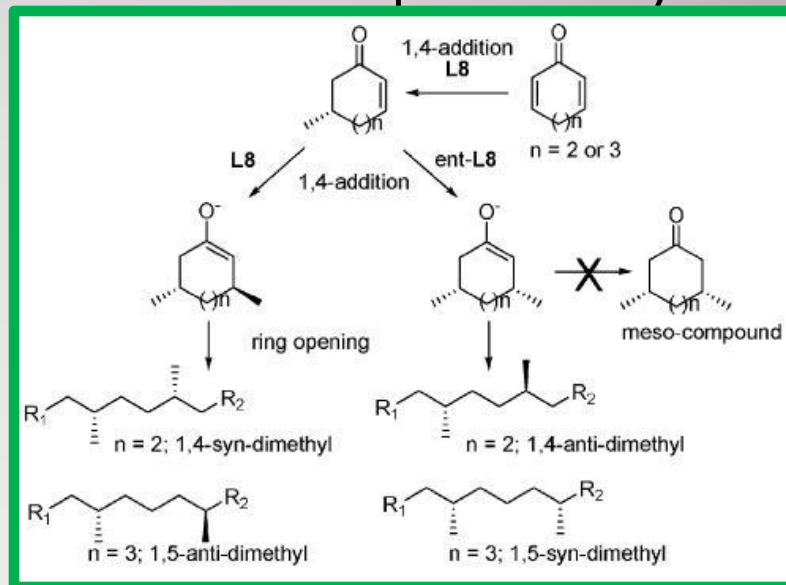
1,4-addition-aldol reactions



Cyclopropanations

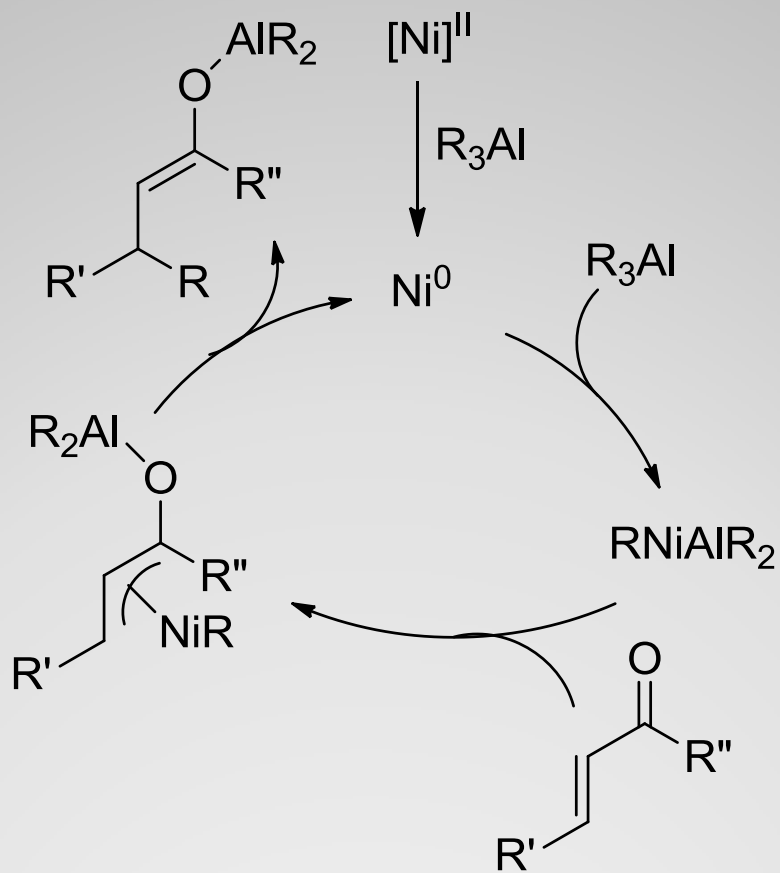


Saturated Isoprenoid Synth

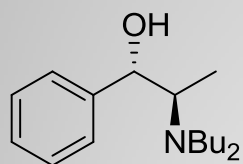
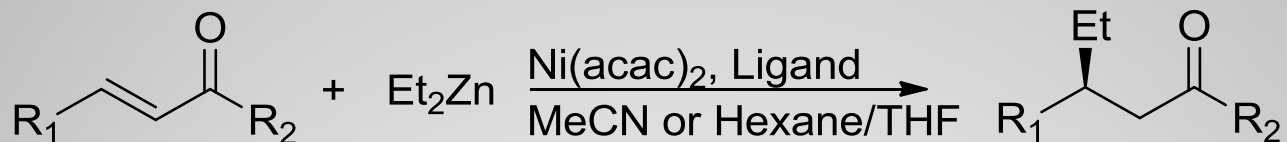
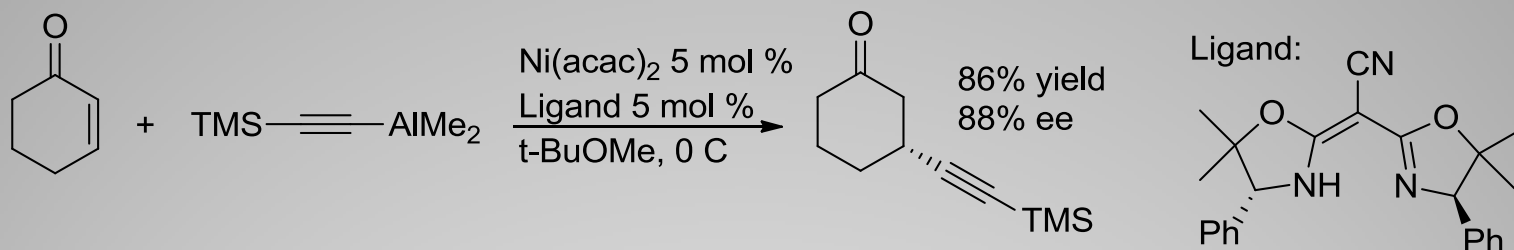


Feringa, B., *J. Am. Chem. Soc.*, **2001**, *123*, 5841.
Alexakis, A., *J. Org. Chem.*, **2002**, *4*, 3835.
Feringa, B., *Chem. Commun.*, **2005**, 1387.

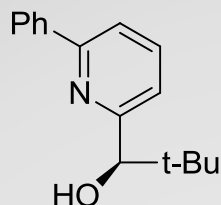
Ni Catalytic Cycle



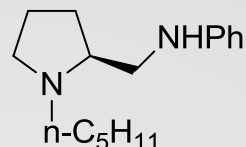
Ni ECA



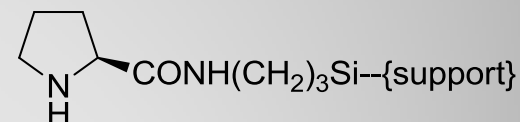
A: 90% ee w/ bipy



B: 86% ee



C: 82% ee



D: 91% ee

Corey, E.J., *Org. Lett.*, **2004**, 6, 3385.

Yokoyama, S., *J. Chem. Soc., Chem. Commun.*, **1989**, 516.

Bolm, C., *Tetrahedron: Asymmetry*, **1991**, 2, 701.

Asami, M., *Chem. Lett.*, **1994**, 297.

Sanchez, F., *Tetrahedron: Asymmetry*, **1992**, 3, 845.

- Cu is well developed and can be used with primarily organozinc, aluminum, or magnesium reagents
- Many functional groups are tolerated
- Primarily used to deliver alkyl groups
- Less is known about Ni, and it tends to be less enantioselective

Summary

Second Row Transition Metals

1 H																	2 He																	
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne																	
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar																	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr																	
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe																	
55 Cs	56 Ba	57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn			
87 Fr	88 Ra	89 Ac	(119)	(120)	(121)	(122)	(123)	(124)	(125)	(126)	(127)	(128)	(129)	(130)	(131)	(132)	(133)	(134)	(135)	(136)	(137)	(138)	(139)	(140)	(141)	(142)	(143)	(144)	(145)	(146)	(147)	(148)	(149)	(150)

LANTHANIDES

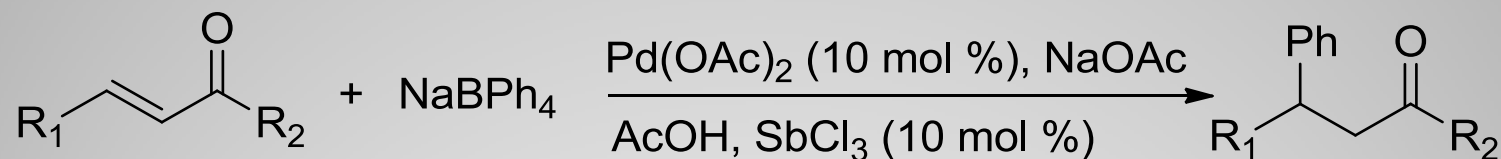
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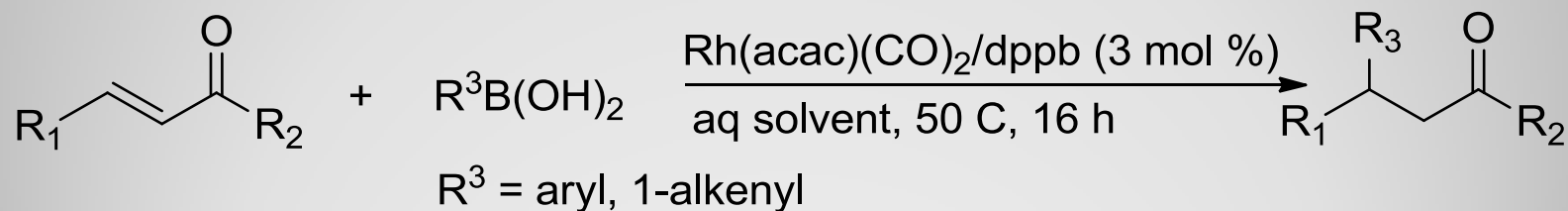
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
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History

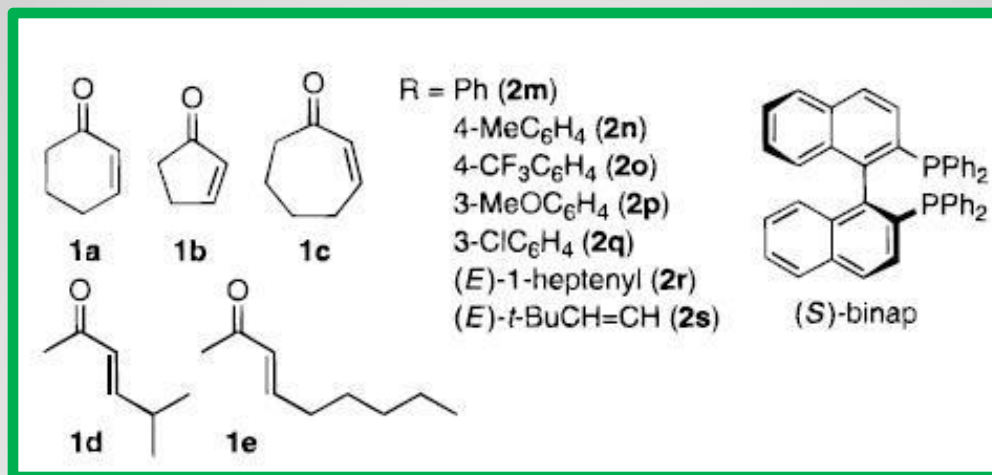
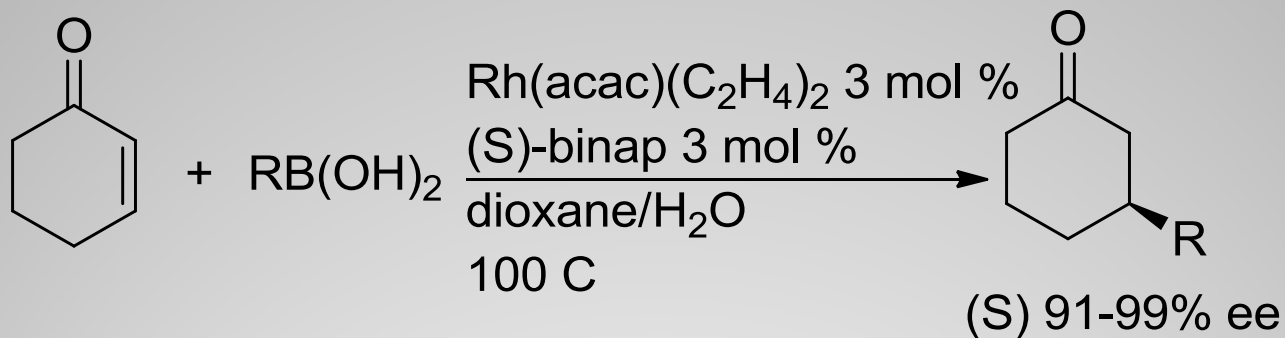
- Uemura-1994



- Miyaura-1997

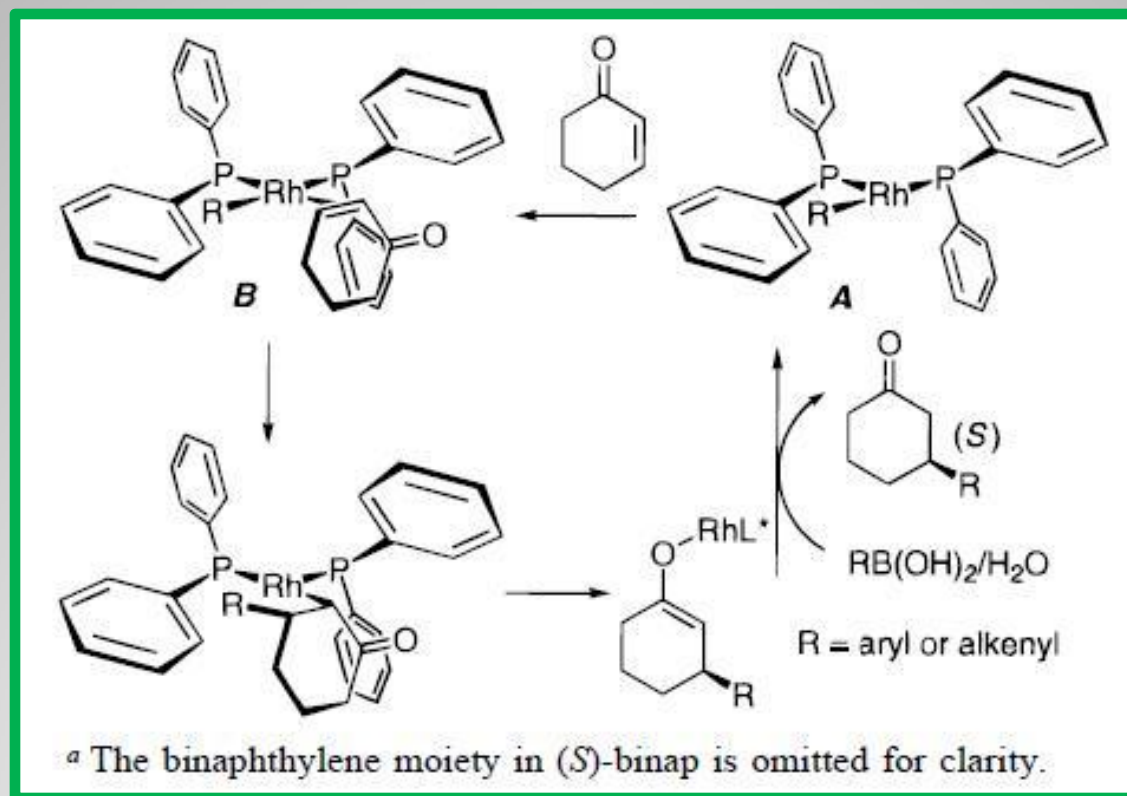


First Example of Enantioselective Rh Catalyzed

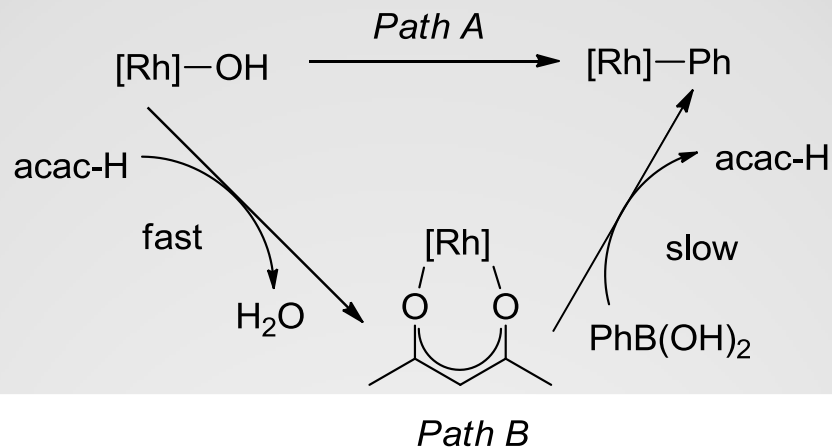
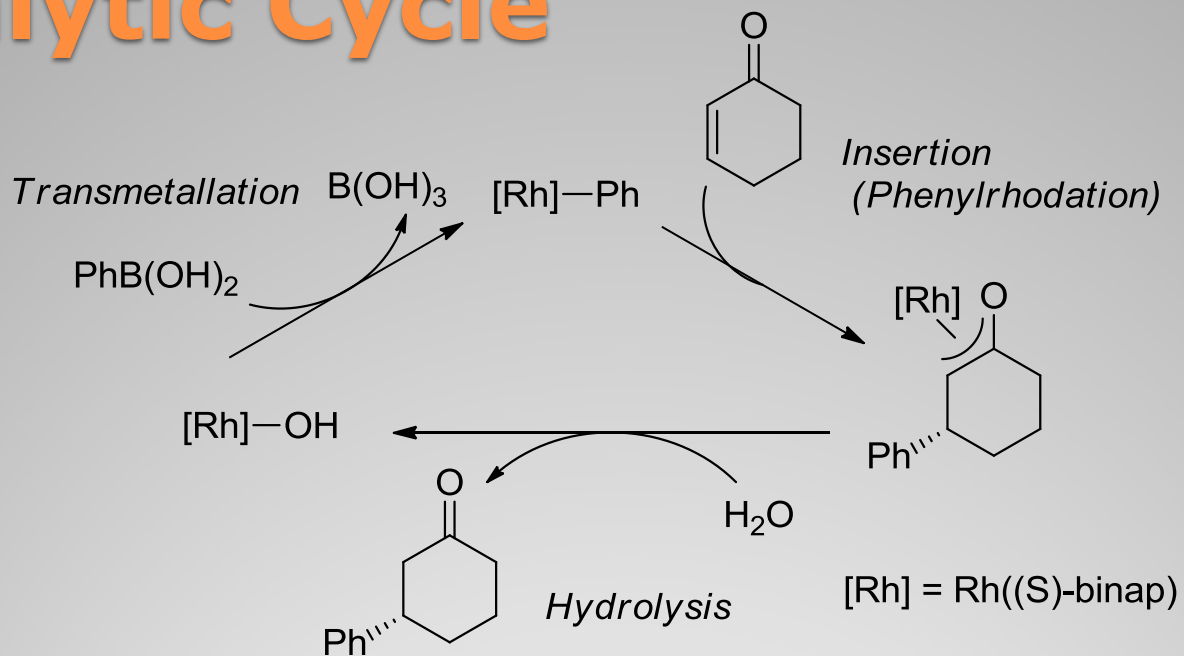


Hiyashi, T., Miyaura, N., *J. Amer. Chem. Soc.*, **1998**, *120*, 5579.

Why Is It Enantioselective?

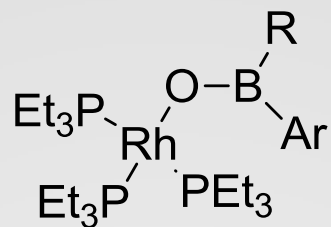
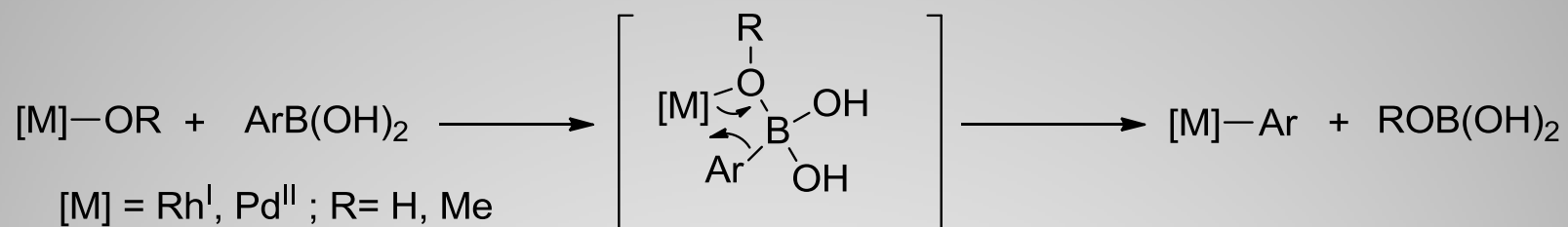


Catalytic Cycle



Hayashi, T., *J. Amer. Chem. Soc.*, **2002**, 124, 5052.

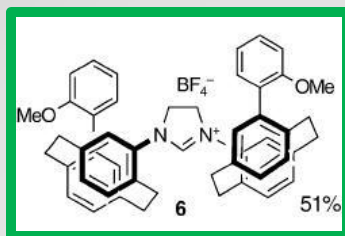
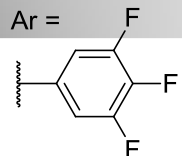
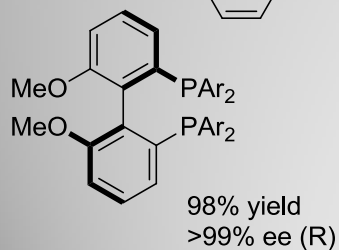
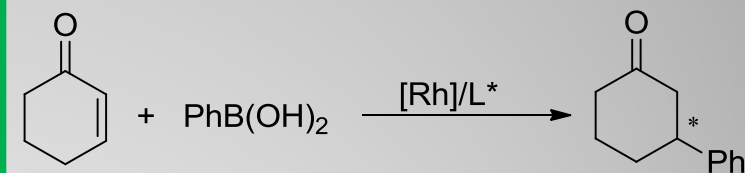
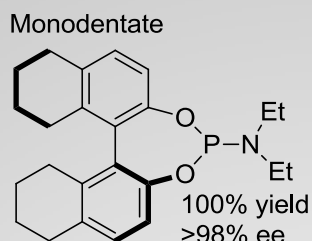
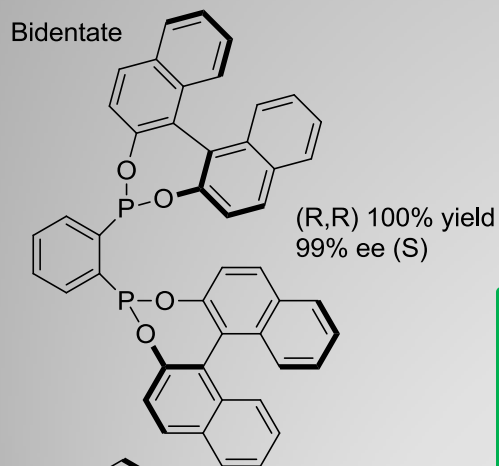
Transmetallation



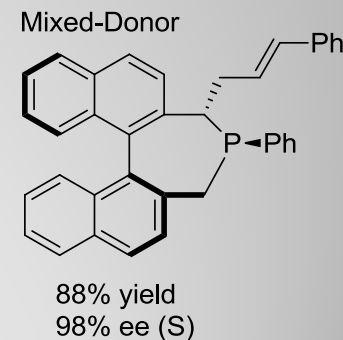
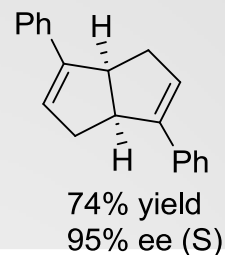
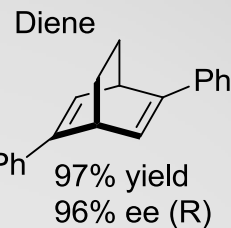
$R = OH, Ar$

Catalyst Precursors/ Organoboron Ligand Scope

- Neutral and cationic Rh species work well
 - $[\text{RhCl}(\text{C}_2\text{H}_4)_2]_2$
 - $[\text{Rh}(\text{nbd})_2]^+\text{BF}_4^-$

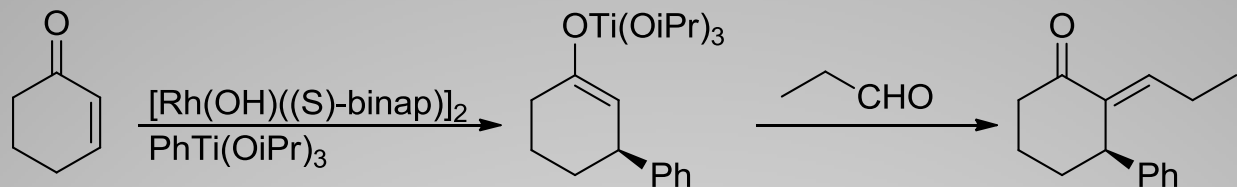


49-99% yield
95->99% ee (R)

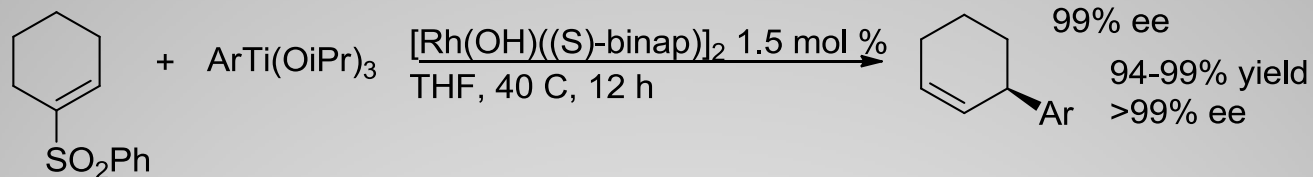


Cordova, A. *Catalytic Asymmetric Conjugate Reactions*, 2010, Wiley-VCH, Weinheim, Germany, 11-19.

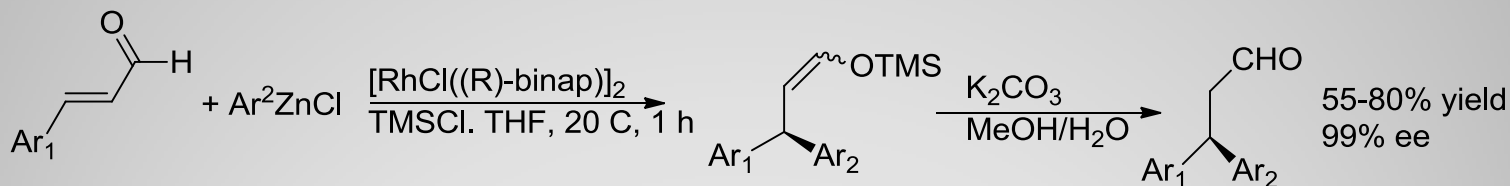
Organotitanium/Organozinc Reagents



45% yield
99% ee

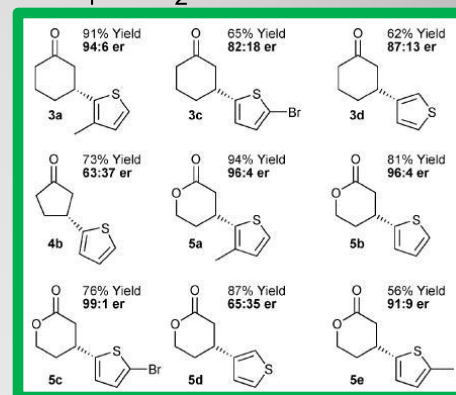
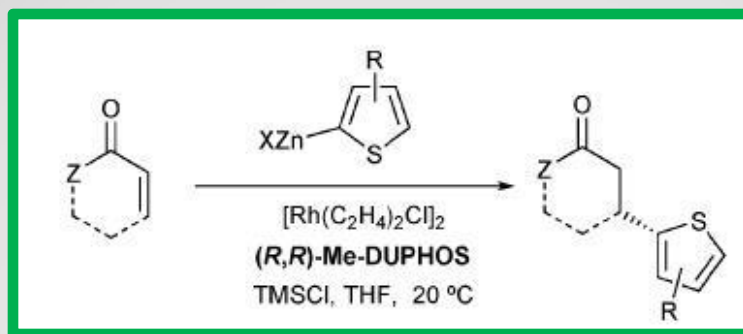
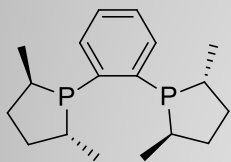


94-99% yield
>99% ee



55-80% yield
99% ee

L: (R,R)-Me-duphos

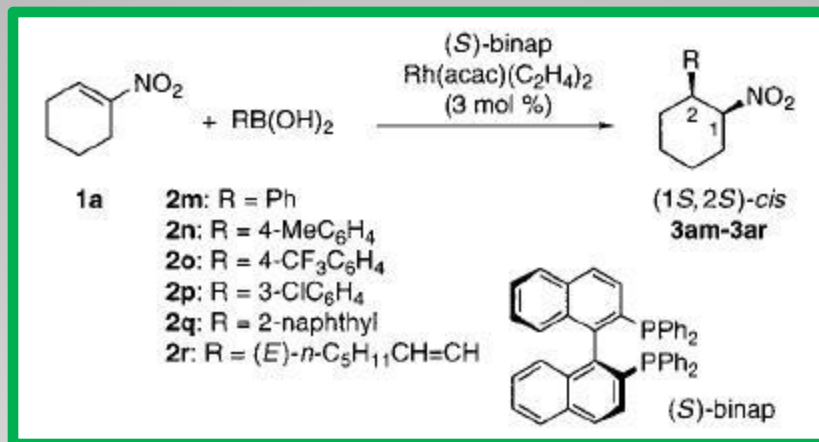


Cordova, A. *Catalytic Asymmetric Conjugate Reactions*, **2010**, Wiley-VCH, Weinheim, Germany, 38, 40.

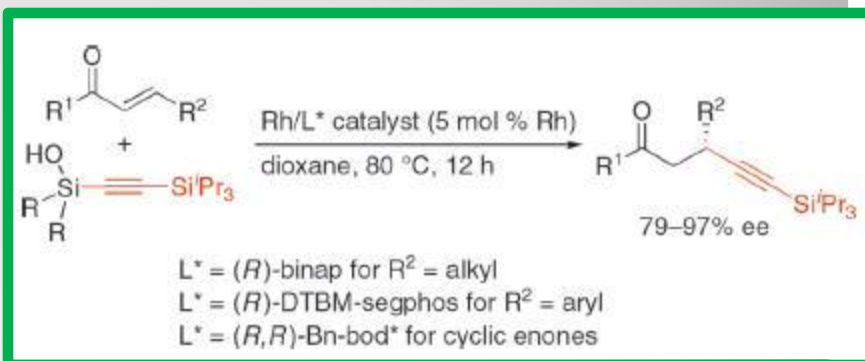
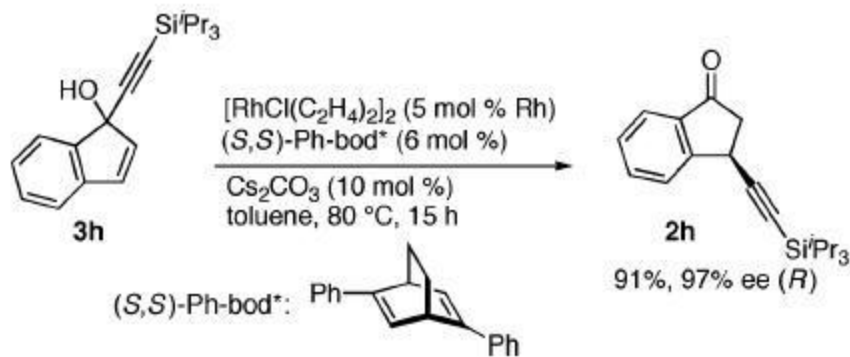
Hayashi, T., *J. Am. Chem. Soc.*, **2003**, 125, 2872.

Frost, C.G., *Chem. Commun.*, **2008**, 3795.

Other Substrates



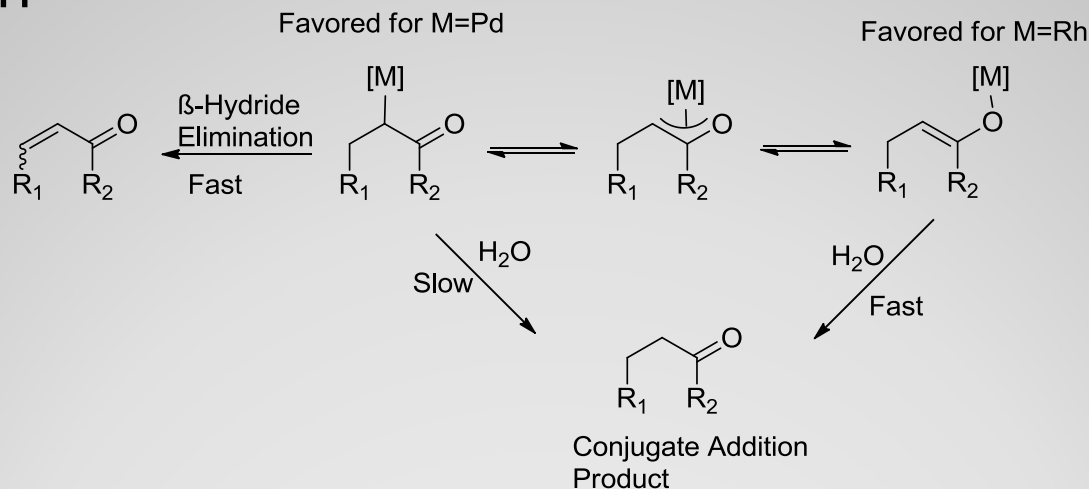
84-85% yield
 98-99% ee
 cis/trans= 88/15-82-12



Hayashi, T., *J. Am. Chem. Soc.*, **2000**, *122*, 10716.
 Hayashi, T., *J. Am. Chem. Soc.*, **2007**, *129*, 14158.
 Hayashi, T., *Org. Lett.*, **2009**, *11*, 3222.

What About Palladium?

- Neutral Pd binds preferentially to carbon, unlike Rh, so β -hydride elimination to Heck-type products is common



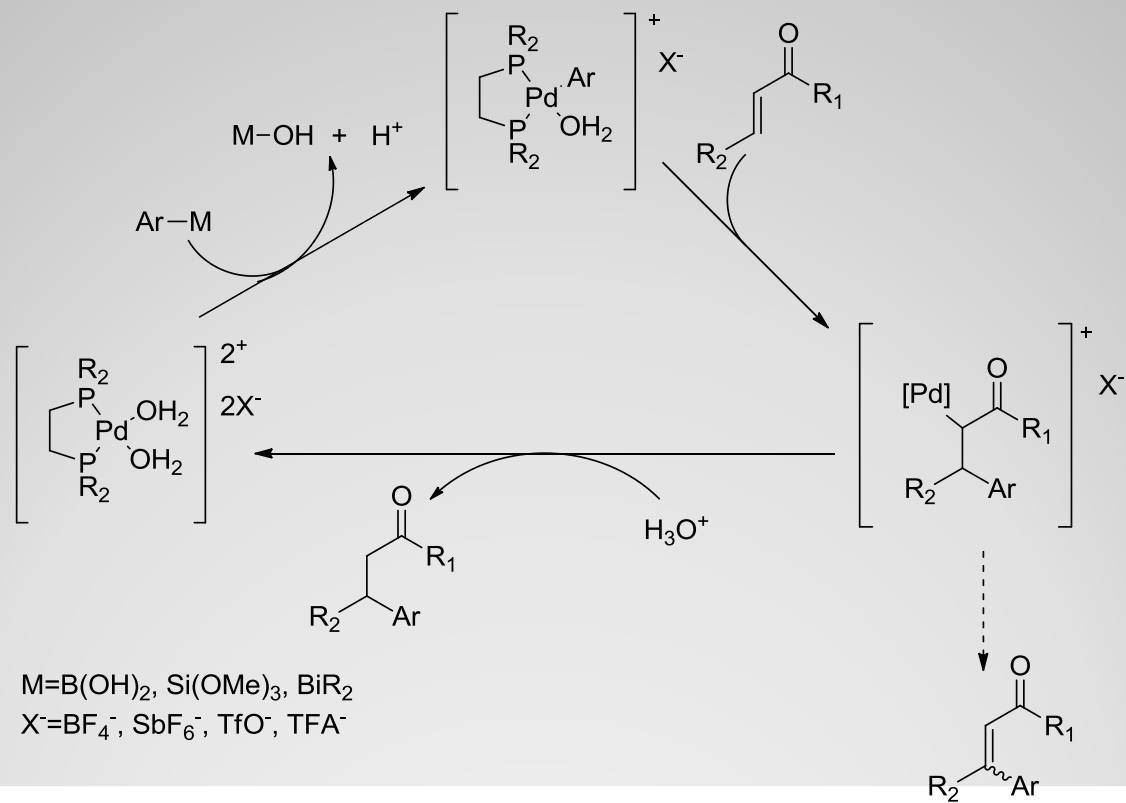
- Cationic Pd can be used to reduce this unwanted side reaction

Cordova, A. *Catalytic Asymmetric Conjugate Reactions*, **2010**, Wiley-VCH, Weinheim, Germany, 145.

Albeniz, A.C., *Organometallics*, **1999**, *18*, 5571.

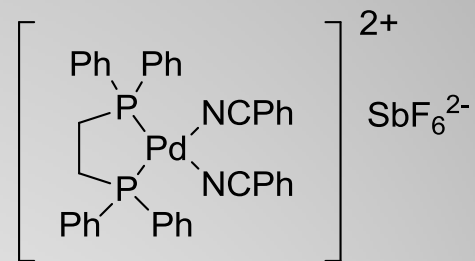
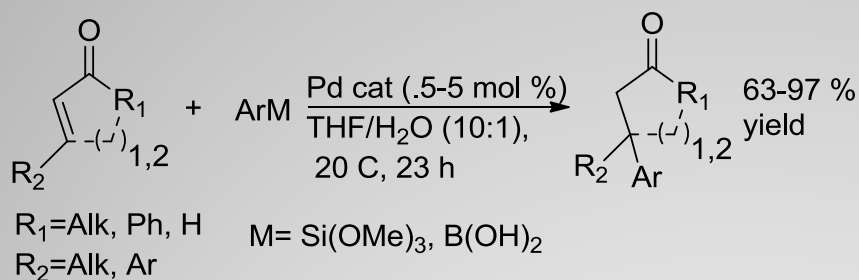
Catalytic Cycle

- Very similar to Rh, except cationic vs neutral species

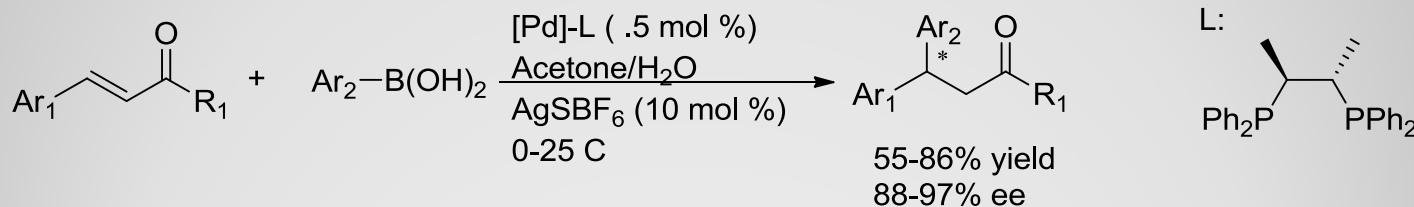


Cationic Pd Complexes

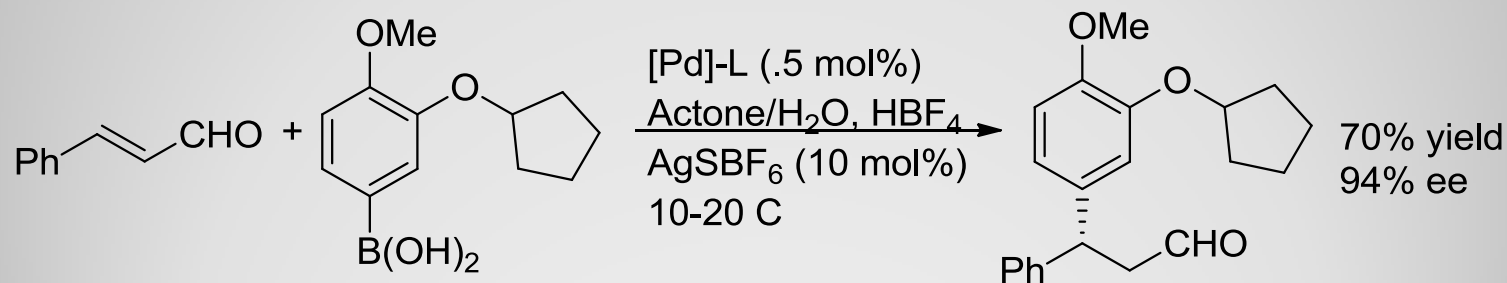
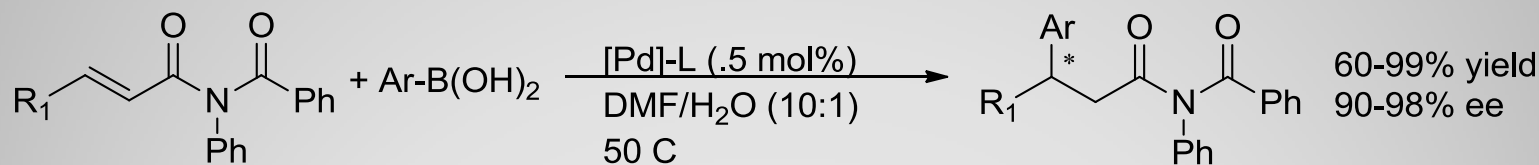
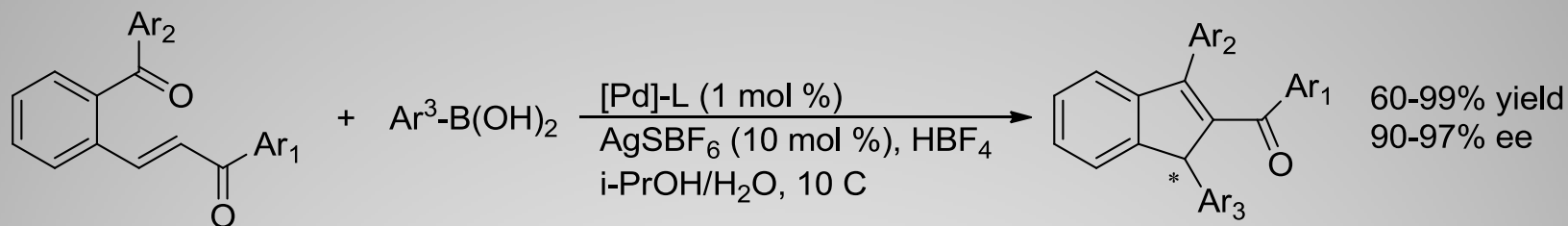
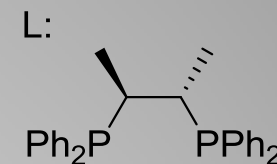
- $[\text{Pd}(\text{dppe})(\text{MeCN})_2](\text{SbF}_6)_2$ was developed by Miyaura.



- Using chiral ligand instead of dppe gave enantioselectivity



Palladium with (*S,S*)-chiraphos



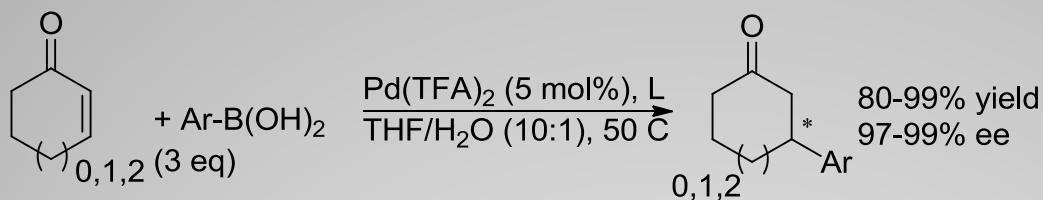
Miyaura, N., *Synlett.*, **2007**, 3055.

Miyaura, N., *Chem. Lett.*, **2007**, 36, 1442.

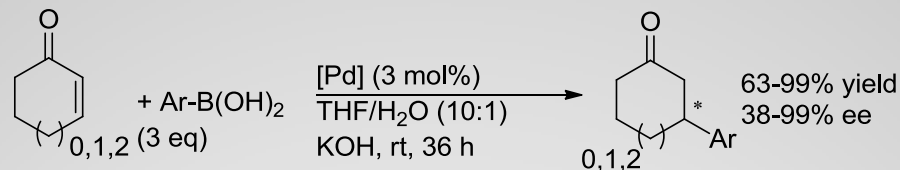
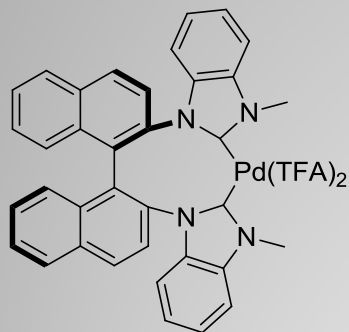
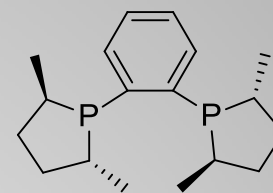
Miyaura, N., *Tetrahedron Lett.*, **2007**, 48, 4007.

Other Pd Catalysts

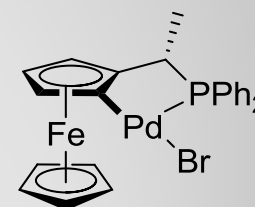
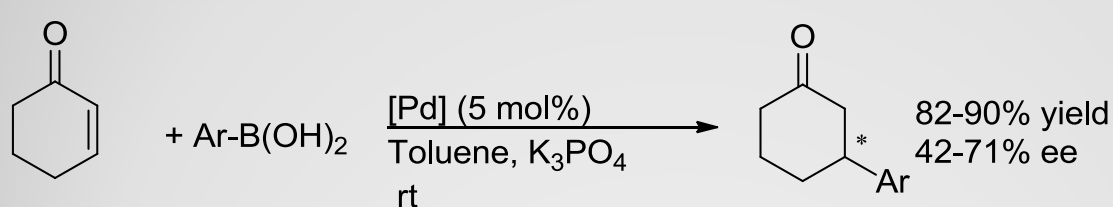
- Pd(TFA)₂



L: (R,R)-Me-duphos



- Palladacycles



Minnaard, A.J., *Org. Lett.*, **2005**, 7, 5309.
Shi, M., *Chem. Eur. J.*, **2008**, 14, 3759.
Ito, Y., *Chem. Lett.*, **2007**, 36, 470.

- Rhodium is expensive, but works well for conjugate addition, with few side reactions
- Palladium is cheaper, but lags behind Rh due to competing β -hydride elimination
- Both work well for aryl and alkenyl groups, but not for sp^3 carbons

Summary

- First and second row TM's complement one another
 - Cu/Ni are good for introducing alkyl groups
 - Rh/Pd are good for introducing aryl/alkenyl groups
- Relatively young field, but growing quickly
- Many catalyst/ligand options
- Tandem reactions allow for great diversity of products

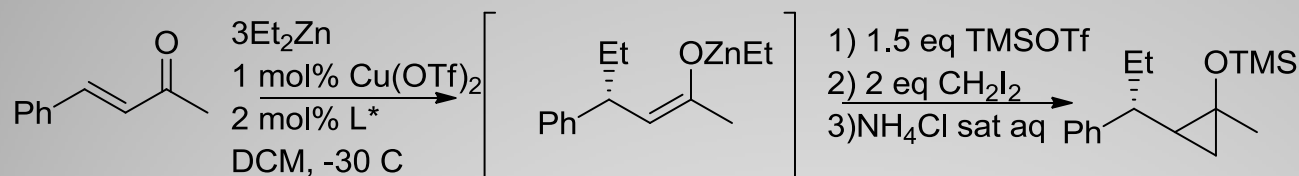
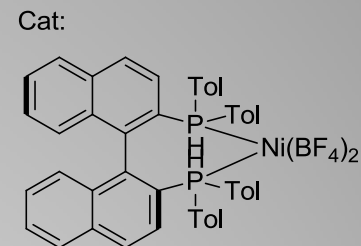
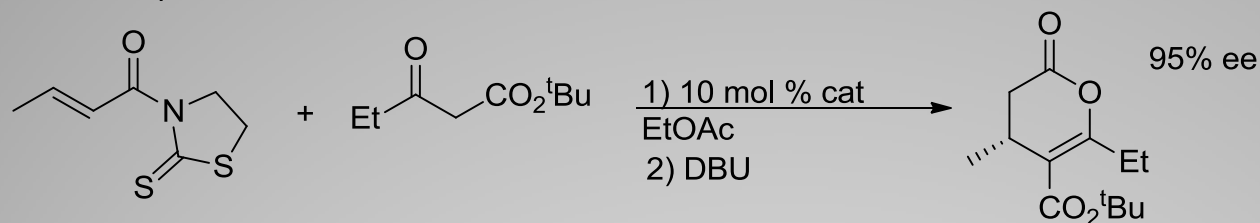
Conclusions

- Thank you for your attention!

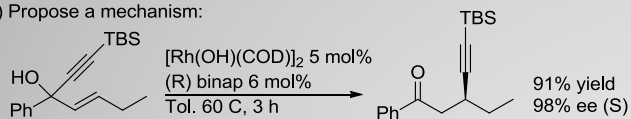


Questions?

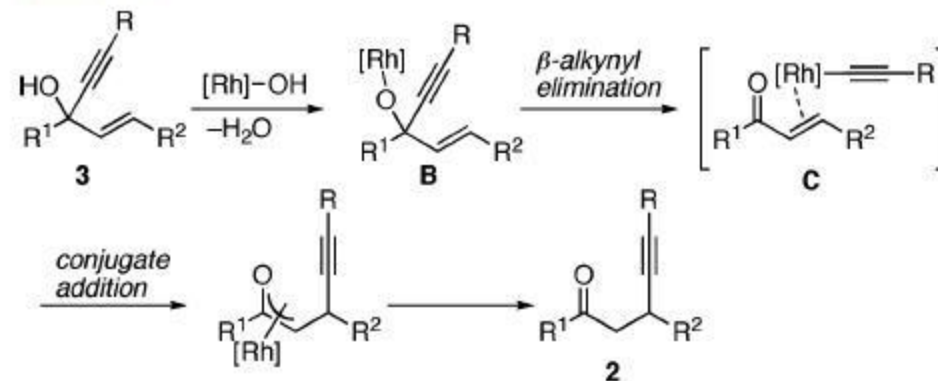
1) Predict the products:



2) Propose a mechanism:

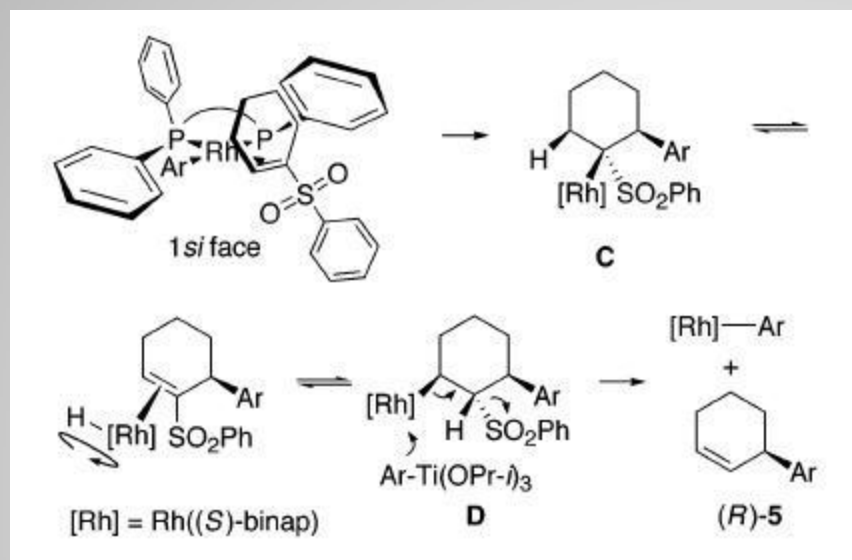
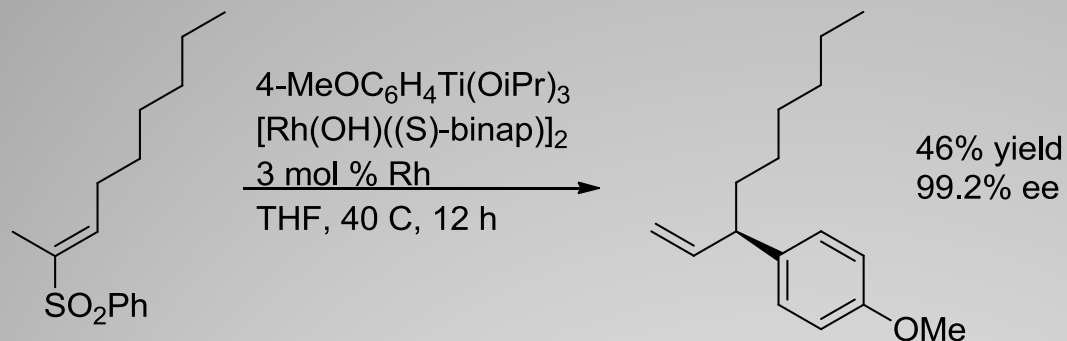


Scheme 2



Question Answers

3) Predict the product and provide a reasonable mechanism:



Question Answers